CALF FLUID THERAPY MADE SIMPLE

Zoe Vogels The Vet Group, PO Box 84 Timboon zvogels@thevetgroup.com.au

A simple explanation of the theories behind fluid therapy. A summary of the latest recommendations. A comparison of products available in Australia. Practical guidelines for fluid therapy in the field. Tips to help farmers act on your advice.

ACID-BASE BALANCE

The pathophysiology of neonatal calf diarrhoea is complex. As the fluid therapy of scouring calves often involves the correction of a lowered blood pH, an understanding of acidbase balance is important.

Acidosis (a decrease in pH) results from an accumulation of hydrogen (H^{*}) ions. For example:

1) Carbon dioxide elimination is inadequate relative to tissue production. Partial pressure of carbon dioxide, pCO_2 , increases and shifts the following equation to the left:

 $\mathsf{HCO}_3^-\mathsf{+}\mathsf{H}^+\leftrightarrow\mathsf{H}_2\mathsf{CO}_3\leftrightarrow\mathsf{CO}_2\mathsf{+}\mathsf{H}_2\mathsf{0}$

2) In body fluids, the sum of all positive ions *must* equal the sum of all negative ions. The difference in charge between the fully dissociated ('strong') cations and anions in the blood influences H⁺ dissociation from body water. When this strong ion difference (SID) becomes more negative (eg. loss of Na⁺, production of lactate⁻), H⁺ concentration rises.

Figure 1 incorporates many references into a pathophysiology flow chart for neonatal calf diarrhoea. For those wanting to read about acid-base balance in more detail, Constable (1999) and Kellum (2005) are recommended.

ORAL REHYDRATION SOLUTIONS

You're just sitting down for lunch (at 3.30pm!) when you're called to the front counter. Farmer Suzie has some scouring calves and wants to know what fluids to give them. Further

questioning reveals that they're all still able to stand and suckle, albeit weakly. You decide oral rehydration is the way to go.

Oral rehydration solutions (ORS) need to contain the following (Smith 2009):

1) Sodium, to draw water into the blood stream to correct dehydration.

2) An excess of strong cations (sodium and potassium) to anions (chloride) to correct strong ion acidosis.

3) Glucose to provide energy and assist sodium absorption. Glycine, acetate, propionate and citrate also assist in sodium absorption.

4) Potassium. Although scouring calves are often hyperkalemic, on correction of dehydration and acidosis there is a total body deficit.

5) An alkalinising agent. This can be bicarbonate, which buffers H⁺, or agents such as acetate, propionate or citrate that consume H⁺ when they're metabolised (Naylor 1996). Acetate and propionate are the agents of choice. They don't alkalinise the small intestine, minimising the risk of any ingested E. coli and Salmonella surviving to cause clinical disease (Constable 2009) and they are also a source of energy (Brooks et al 1996). don't interfere Thev with milk digestion.

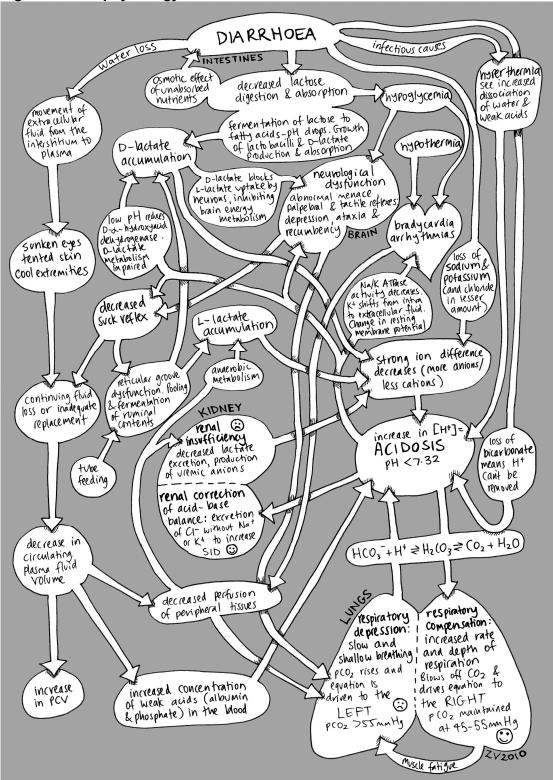


Figure 1 Pathophysiology of Neonatal Calf Diarrhoea

Incorporated references: Abeysekara et al 2007, Constable 1999, Ewaschuk et al 2004, Ewaschuk, Naylor and Zello 2005, Kellum 2005 and Lorenz 2003.

Product, Company and mixing rate		ing rate	Sodium	Potassium	Chloride	SID [%]	A	kalinizing agent	Glucose:Na^	Osmolarity~	Energy*
Recommended values (Smith		th 2009)	90-130 mmol/L	10-30 mmol/L	40-80 mmol/L	60-80 mEq/L	50-80 mmol/L Acetate, propionate		1:1 to 3:1	500-600 mOsm/L	MJ/L
Bovelyte	Provimi	90g/2L	112	18	46	84	84	Bicarbonate (66) Citrate (18)	1.7	435	0.53
Diarrest sachets	Virbac	248.5g/2L	148	31	103	76	60	Bicarbonate (18) Citrate (12) Acetate (30)	1.2	526	1.26
Hydrate liquid concentrate	Bomac	160ml/2L	76	15	74	17	2	Propionate	2.3	359	0.38
Hydrate sachets	Bomac	68.5g/2L	76	15	74	17	2	Propionate	2.2	347	0.35
Lactolyte sachets	Virbac	90g/2L	79	28	55	52	47	Acetate (30) Propionate (17)	1.3	328	0.64
Lectade sachets	Jurox	64g/2L	73	16	73	16	4	Citrate	2.2	344	0.35
Lite-Scour-Diet	Agvantage	20g/2L	12	3	10	5	5	Bicarbonate	4.3	79	0.14
Maxi-Trans	Virbac	20g/2L	12	3	10	5	5	Bicarbonate	4.3	79	0.14
Mega-Lyte	Sykes	100g/2L	88	0	59	29	29	Bicarbonate	2.5	399	0.62
Mega-Lyte Plus	Sykes	60g/1.5L	82	14	46	50	50	Bicarbonate (36) Citrate (14)	1.1	291	0.26
Pronto	Dasco	100g/2L	84	0	57	27	27	Bicarbonate	2.5	383	0.60
Res-Q	DeLaval	100g/2L	173	0	114	59	59	Bicarbonate	2.6	791	1.25
Resus	Dasco	100g/2L	84	0	57	27	27	Bicarbonate	2.5	383	0.60
Scour-Ade sachets	Sykes	248.5g/2L	148	31	103	76	60	Bicarbonate (18) Citrate (12) Acetate (30)	1.2	526	1.26
Scourlyte	Novartis	110g/2L	89	19	76	32	32	Citrate	1.8	356	0.47
Veanalyte Electroguard	Veanavite	60g/1.5L	79	14	46	47	47	Bicarbonate (34) Citrate (13)	1.1	285	0.26
Vy'Trate Liquid Concentrate	Jurox	160ml/2L	73	31	73	31	5	Citrate	2.3	373	0.35
Vy'Trate Sachets	Jurox	64g/2L	73	16	73	16	4	Citrate	2.2	344	0.35
Zoe's recipe 11		110g/2L	124	13	65	73	73	Acetate	1.8	503	0.70

Table 1: Composition of oral rehydration solutions available in Australia[#]

Composition calculated using Microsoft Excel. Only products whose mixing rates are known are included (APVMA 2010, MIMS 2010).

[%] SID = [Na] + [K] - [Cl]

A Glycine (in Hydrate, Lectade and Vy-Trate) is included in the Glucose:Na ratio. It is assumed that the lactose in Diarrest, Lactolyte and Scour-Ade was totally digested.

[~] Osmolarity >700mOsm/L increases the risk of osmotic diarrhoea and ileus (Smith 2009).

* Glucose, lactose & rice flour $\rightarrow \sim 16$ kJ/g, acetate $\rightarrow 0.9$ kJ/mmol, citrate $\rightarrow 2.1$ kJ/mmol, propionate $\rightarrow 1.5$ kJ/mmol (Brooks et al 1996). A 45kg calf on a milk-only diet has a maintenance requirement of 7.26MJ/day (NRC 2001). Excluding Lite-Scour-Diet and Maxi-Trans, these products fed at 4L/day contain $\sim 15-70\%$ of this requirement.

COMMERCIAL PRODUCTS

Table 1 contains the mixing instructions and composition of oral electrolvte products available in When compared with Australia. recommended values (Smith 2009), no product is 'ideal'. But, looking at the components that affect acid-base balance (alkalinising ability and strong ion difference and SID), several are good: Bovelvte, Diarrest, Lactolvte, Mega-Lyte Plus, Res-Q and Scour-Ade. The cost price for farmers per 2L dose is roughly 70c-\$2 for powders available in buckets, \$3 for liquid concentrates and \$4-7 for sachets.

Lactolyte's main source of energy is lactose (Diarrest and Scour-Ade contain both lactose and glucose). Lactose is hydrolysed at the intestinal mucosa by lactase to form glucose and galactose. Lactase activity can be halved with rotavirus infection and decreased even further in combined rotavirus/E. coli infections (Tzipori et al 1983). Calves that are unable to turn lactose into glucose and galactose may have insufficient glucose to assist sodium and water absorption. The risk of intestinal fermentation and D-lactate production will also be increased (see Figure 1).

A homemade recipe that meets all the recommendations can be made:

12g sodium acetate anhydrous2g potassium chloride6g sodium chloride90g dextrose monohydrate2L water

To make 2.2kg buckets you need:

240g sodium acetate anhydrous40g potassium chloride120g sodium chloride1800g dextrose monohydrate

Batches of 10–20kg can be blended easily in a drum mixer and divided out (C. Wood, 2010, pers. comm., 21 Jan). Including labour, the cost of this homemade recipe is comparable to the cheaper commercial products. Kept clean and dry, the electrolyte mixture should remain stable if combined with water just prior to use (J. Fairweather, Sypharma Pty Ltd, 2010, pers. comm., March 10).

Choice of ORS will need to take into account convenience (both for the vet and the farmer), composition, cost and intravenous fluid regimes (see later).

HOW MUCH TO GIVE?

Daily maintenance fluid requirements for calves range from 5% body weight to 10% in hot weather. Table 2 assumes 70% absorption efficiency and ongoing losses of 2L per day while scouring (Naylor 1996). Any preexisting dehydration would need to be treated with additional fluid. It is important to note that the dosage instructions for many ORS products do not supply a scouring calf's daily fluid requirement (APVMA 2010, MIMS 2010).

Table 2: Daily fluid requirements forscouring calves

BW	Feeds/day	Feeds/day
(kg)	Cold weather	Hot weather
30	4 x 1.3 L	4 x 1.8 L
40	4 x 1.4 L	4 x 2.2 L
50	4 x 1.6 L	4 x 2.5 L
60	4 x 1.8 L	4 x 2.9 L

Studies have shown that diarrhoeic calves fed milk and ORS gain weight throughout treatment and their diarrhoea not worsened or is prolonged (Garthwaite et al 1994, Heath et al 1989). Maintain calves on their normal milk diet, with additional ORS feeds. However, if calves become depressed or refuse to suckle (i.e. D-lactate acidosis is suspected), withhold a milk feed and substitute with ORS (Ewaschuk, Naylor & Zello 2005, Smith 2009).

Calves should have access to ad lib water.

INTRAVENOUS FLUID THERAPY

You've just finished a rotten calving. As you're about to get in your car and drive off for a much needed shower, farmer Bob says "While you're here..." and takes you to the calf shed where a week old scouring calf is lying comatose on the ground."

A calf such as this is likely to be suffering severe acidosis. The ideal treatment is intravenous administration of isotonic (1.3%) sodium bicarbonate (NaHCO₃). It provides HCO_3^- to buffer H⁺ and prevent further decrease in pH. It also provides Na⁺ ions to increase the SID. The fluid expands plasma volume and assists in the excretion of D-lactate (Berchtold et al 2005).

However, 1.3% NaHCO₃ is unavailable commercially. Bicarbonate solutions are unstable if stored in permeable plastic due to atmospheric CO₂ loss. They must be stored in glass and infused promptly (Morgan 2005). A sterile 1.3% NaHCO₃ product is under development (J. Fairweather, Sypharma Pty Ltd, 2010, pers. comm., February 10) and it is hoped to be available in the near future.

In the meantime, 13g of NaHCO₃ can be added to 1L water for injection or, if unavailable, 1L 0.9% NaCl. Place 10ml (2 cooking teaspoons) into a 20ml syringe, dissolve in some of the water or NaCl and squirt back into the bag. An alternative (and more expensive!) option is to add 155ml 8.4% NaHCO₃ to 845ml of your chosen fluid.

Tables 3, 4 and 5 take into account the degree of acidosis predicted by a recumbent calf's clinical signs and age (Naylor 1989) and calculate how many litres of 1.3% NaHCO₃ the calf needs.

Table	3:	Calf ≤8	davs	old
IUNIC	۰.		auyo	ora

30	40	50	60
1.4L	1.8L	2.3L	2.8L
1.5L	2.0L	2.5L	3.0L
	1.4		

Table 4: Calf >8 days old

BW (kg)	30	40	50	60
Sternal	1.8L	2.5L	3.1L	3.7L
Lateral	2.3L	3.1L	3.8L	4.6L

Table 5:	Tables	3 and 4	simplified
----------	--------	---------	------------

Calf size	≤40kg	>40kg
≤8 days, recumbent	2L	3L
>8 days, sternal recumbency	2L	3L
>8 days, lateral recumbency	3L	4L

Maximum recommended fluid rates are 80ml/kg/hour (4L/h in a 50kg calf). Unless other conditions, such as septicemia, are evident, calves should respond quickly. They can be switched to ORS once they can stand and suckle. Calves still depressed the next day can be given additional bicarbonate (Naylor 1996).

Jugular catheterisation:

- 1) Prepare 1.3% sodium bicarbonate
- 2) Fill 20ml syringe with 10ml fluid
- 3) Warm fluids and attach giving set
- 4) Shave skin with razor and prep with iodine
- 5) Nick the skin with a scalpel
- Insert 18–gauge 32mm catheter. When you see blood, remove stilet, attach the syringe and push in fluid while advancing the catheter
- 7) Attach giving set and superglue catheter to the skin, with a loop of the giving set glued to the calf's ear. Set drip going and ensure calf isn't able to wander once it is standing.

A technique for auricular catheterisation is described in detail (Berchtold 2009) but only allows for fluid administration rates of ~1-2L/h (Radostits et al 2007).

Frequently calves have clinical signs of severe acidosis with minimal dehydration. At other times, additional fluid is required to correct a concurrent moderate to severe dehydration. Lactated and acetated Ringer's solutions can be used. Compared to other commercially available isotonic solutions, their effective SID reduces the chance of an acid-base disturbance (Constable 2003, Morgan 2005).

Eyeball recession and skin tenting (Table 6) provide the best assessment of dehydration (Smith 2009).

Table 6: Volume of fluid required tocorrect degree of dehydration

Dehydration	8%	10%	12%
Skin tent (sec)*	2	5	10
Eyeball recession	4mm	6mm	8mm
30 kg	2.4L	3.0L	3.6L
40 kg	3.2L	4.0L	4.8L
50 kg	4.0L	5.0L	6.0L
60 kg	4.8L	6.0L	7.2L

*Pinch a fold of neck skin, rotate it 90 degrees and time how long it takes to return to normal

For example, a 10 day old, 50kg calf in lateral recumbency and 12% dehydrated would receive 4L of 1.3% bicarbonate and another 2 L of lactated or acetated Ringer's.

WHEN TIME IS OF THE ESSENCE...

You're taking a drive with a beef farmer, checking out his scouring calves, when you see a recumbent calf in a paddock. You need to treat it quickly without getting gored by its dam. What are some options?

Other intravenous fluids

Commercial colloid solutions contain high molecular weight glucose polymers. By increasing plasma oncotic pressure they draw fluid from the interstitium to the intravascular space (Constable 2003).

In calves, administration of 6% dextran-70 combined with hypertonic saline and ORS increases cardiac output and plasma volume and decreases blood lactate concentration and hematocrit (Constable et al, 1996). The hypertonic saline draws

water into the blood and the hyperoncontic dextran-70 keeps it there. To make this solution, add 31.6g NaCl to a 500ml bag of 6% dextran-70 in 0.9% saline (Constable Refrigerate if not using 2003). immediately. Administer at a rate of 4-5ml/kg, over 4-5 minutes, and tube feed 55ml/kg ORS. The ORS should have a low osmolarity (preferably 300-312mOsm/L) and have bicarbonate as its alkalinising agent to increase blood pH rapidly (Naylor 1996).

Pentastarch and Voluven are other commercial colloids available as 6% solutions. Theoretically they could be used in the same way.

Intraperitoneal fluids

Intraperitoneal (IP) fluids have the potential to give colloids or isotonic fluids easily (Constable 2003). A catheter or needle is inserted 10cm ventral to the lateral processes of the lumbar vertebrae, midway between the last rib and the tuber coxae (Radostits 2007). The ability to use larger bore catheters allows much faster rates of fluid administration than with IV fluids. There have not been many studies on the use of IP fluids in calves. There are questions about altered peritoneal absorption and risk of adhesions (Constable 2003) and further studies would be useful.

GIVING ADVICE

Each farmer and situation is different. Here are some strategies to help get your message through – you just need to TWEET:

1) Have no more than THREE main points, for example: 'Colostrum, electrolytes, isolation'.

2) WRITE instructions down. This ensures that instructions aren't forgotten, especially if there's been some time since the last scours outbreak. Your advice is then available for other people as well. Tailor-make handouts for your clinic and have them photocopied and ready to be filled in and left on-farm.

3) EMPATHISE. Having sick calves is an extra stress for farmers. How you impart your advice needs to take this into account: 'I know it's a big ask, but can you manage an extra 2 feeds of electrolytes a day...'

4) EXPLAIN your recommendations using analogies that will mean something to the farmer: 'Not having extra electrolytes is like you playing a whole footy match without water'.

5) TELEPHONE to see how things are going. Adapt any of your recommendations that aren't working.

SUMMARY

Neonatal calf diarrhoea is a complex disease process. However, cattle practice in Australia needs simple, practical fluid therapy. Calves able to stand and suckle should be treated with ORS with adequate an alkalinising ability and SID. Recumbent calves with acidosis ideally need 1.3% bicarbonate IV. Combined hypertonic/colloid solutions IV allows faster expansion of plasma volume, intraperitoneal isotonic fluids or colloids have the potential to allow rapid infusion of fluids.

References:

Abeysekara S, Naylor JM, Wassef AWA, Isak U, Zello GA (2007) D-lactic acid-induced neurotoxicity in a calf model Am J Physiol Endocrinol Metab 293:558-565

Australia Pesticides and Veterinary Medicines Authority (APVMA). Public Chemical Registration Information System Available: <u>http://services.apvma.gov.au/Pubcris</u> <u>WebClient/welcome.do</u> [Accessed March 2010]

Berchtold JF, Constable PD, Smith GW, Mathur SM Morin DE, Tranquilli WJ (2005) Effects of intravenous hyperosmotic sodium bicarbonate on arterial and cerebrospinal fluid acidbase status and cardiovascular function in calves with experimentally induced respiratory and strong ion acidosis J Vet Intern Med 19: 240-251

Berchtold J (2009) Treatment of calf diarrhea: intravenous fluid therapy Vet Clin Food Anim 25:73-99

Brooks HW, White DG, Wagstaff AJ, Michell AR (1996) Evaluation of a nutritive oral rehydration solution for the treatment of calf diarrhoea Br Vet J 152: 699-708

Constable PD, Gohar HM, Morin DE, Thurmon JC (1996) Use of hypertonic saline-dextran solution to resuscitate hypovolemic calves with diarrhoea. Am J Vet Res 57, 1: 97-104

Constable PD (1999) Clinical assessment of acid-base status Vet Clin Food Anim 15:447-471

Constable PD (2003) Fluid and electrolyte therapy in ruminants Vet Clin Food Anim 19:557-597

Constable PD (2009) Treatment of calf diarrhoea: antimicrobial and ancillary treatments Vet Clin Food Anim 25: 101-120

Ewaschuk JB, Naylor JM, Palmer R, Whiting SJ, Zello GA (2004) D-lactate production and excretion in diarrheic calves. J Vet Intern Med 18:744-747

Ewaschuk JB, Naylor JM, Zello GA (2005) D-lactate in human and ruminant metabolism J Nutr 135:1619-1625

Garthwaite BD, Drackley JK, McCoy GC, Jaster EH (1994) While milk and oral rehydration solution for calves with diarrhea of spontaneous origin J Dairy Sci 77:835-843

Heath SE, Naylor JM, Guedo BL, Petrie L, Rousseaux CG, Radostits OM (1989) The effects of feeding milk to diarrheic calves supplemented with oral electrolytes Can J Vet Res 53:477-485

Kellum JA (2005) Determinants of plasma acid-base balance Crit Care Clin 21:329-346

Lorenz I (2003) Investigations on the influence of serum D-lactate levels on clinical signs in calves with metabolic acidosis The Veterinary Journal 168:323-327

MIMS Australia (2010) IVS Annual 22nd edition UBM Medica Australia Pty Ltd

Morgan TJ (2005) Clinical review: the meaning of acid-base abnormalities in the intensive care uni-effects of fluid administration Critical Care 9:204-211

National Research Council (NRC) (2001) Nutrient Requirements of Dairy Cattle 7th Revised edition, Chapter 10: Nutrient Requirements of the Young Calf Naylor JM (1989) A retrospective study of the relationship between clinical signs and severity of acidosis in diarrheic calves Can Vet J 30: 577-580

Naylor, JM (1996) Neonatal ruminant diarrhea In: Large Animal Medicine, Smith Ed. 396-417. 2nd ed. Mosby– Year Book Inc

Radostits OM, Gay CC, Hinchcliff KW, Constable PD (2007) Veterinary Medicine, 10th ed. Elsevier Ltd.

Smith GW (2009) Treatment of calf diarrhea: oral fluid therapy Vet Clin Food Anim 25:55-72

Tzipori S, Smith M, Halpin C, Makin T and Krautil F (1983) Intestinal changes associated with rotavirus and enterotoxigenic Escherichia coli infection in calves Vet Microbiol 8:35-43